Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

### 5-(4-Bromophenyl)-2-(3,4-methylenedioxyphenyl)-3-methylsulfanyl-1-benzofuran

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Received 4 September 2009; accepted 21 September 2009

Key indicators: single-crystal X-ray study; T = 173 K; mean  $\sigma$ (C–C) = 0.012 Å; R factor = 0.061; wR factor = 0.150; data-to-parameter ratio = 12.9.

The title compound, C<sub>22</sub>H<sub>15</sub>BrO<sub>3</sub>S, crystallizes with four molecules in the asymmetric unit. The 4-bromophenyl rings are rotated out of the benzofuran planes, with dihedral angles for the four molecules of 20.8 (2), 17.8 (2), 23.5 (4) and 23.9 (4)°. The dihedral angles between the 3,4-methylenedioxyphenyl ring and the benzofuran plane are 13.5 (2), 7.1 (2), 18.6 (3) and 14.2 (3) $^{\circ}$  in the four molecules. The crystal structure is stabilized by weak nonclassical intermolecular C-H···O hydrogen bonds. The crystal structure also exhibits intermolecular aromatic  $\pi$ - $\pi$  interactions between the benzene and furan rings and between the 4-bromophenyl and 3,4-methylenedioxyphenyl rings from molecules of the same type; the centroid-centroid distances are 3.92 (1) and 3.79 (1), 3.91 (1), 3.77 (1) and 3.77 (1), and 3.79 (1) and 3.75(1)Å in the four molecules.

#### **Related literature**

For the crystal structures of similar 3-methylsulfanyl-2-phenyl-1-benzofuran derivatives, see: Choi, Seo et al. (2006); Choi, Woo et al. (2006). For natural products of benzofuran ring systems, see: Akgul & Anil (2003); von Reuss & König (2004).



#### **Experimental**

### Crystal data

α β

$C_{22}H_{15}BrO_3S$	$\gamma = 68.0552 \ (9)^{\circ}$
$M_r = 439.31$	V = 3592.7 (4) Å <sup>3</sup>
Triclinic, P1	Z = 8
a = 12.3757 (8) Å	Mo $K\alpha$ radiation
b = 16.067 (1)  Å	$\mu = 2.43 \text{ mm}^{-1}$
c = 19.587 (1)  Å	$T = 173  { m K}$
$\alpha = 84.078 \ (1)^{\circ}$	$0.24 \times 0.16 \times 0.12 \text{ mm}$
$\beta = 88.6573 \ (9)^{\circ}$	

#### Data collection

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.061$	
$wR(F^2) = 0.150$	
S = 1.06	
12562 reflections	

### Table 1

Hydrogen-bond geometry (A	A, °]	).
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$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
C12-H12A···O4	0.99	2.58	3.50(1)	154
$C21-H21\cdots O6^{i}$	0.95	2.51	3.37 (1)	150
$C57 - H57A \cdots O10^{ii}$	0.99	2.50	3.32 (1)	141
C65-H65···O11 <sup>iii</sup>	0.95	2.58	3.47 (1)	156
$C79-H79B\cdots O7^{iv}$	0.99	2.56	3.26 (1)	128

26887 measured reflections 12562 independent reflections

 $R_{\rm int} = 0.094$ 

973 parameters

 $\Delta \rho_{\rm max} = 0.53 \ {\rm e} \ {\rm \AA}^-$ 

 $\Delta \rho_{\rm min} = -0.73 \text{ e} \text{ Å}^{-3}$ 

6573 reflections with  $I > 2\sigma(I)$ 

H-atom parameters constrained

Symmetry codes: (i) -x + 2, -y + 1, -z; (ii) -x + 1, -y + 1, -z + 1; (iii) x, y + 1, z; (iv) -x, -y + 1, -z + 1.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997) and DIAMOND (Brandenburg, 1998); software used to prepare material for publication: SHELXL97.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PB2008).

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Acta Cryst. (2009). E65, o2530 [doi:10.1107/S1600536809038148]

### 5-(4-Bromophenyl)-2-(3,4-methylenedioxyphenyl)-3-methylsulfanyl-1-benzofuran

#### H. D. Choi, P. J. Seo, B. W. Son and U. Lee

#### Comment

Molecules containing benzofuran skeleton constitute a major group of naturally-occurring compounds that are of considerable interest because of their biological activities (Akgul & Anil, 2003; von Reuss & König, 2004). As a part of our ongoing studies of the effect of side chain substituents on the solid state structures of 3-methylsulfanyl-2-phenyl-1-benzofuran analogues (Choi, Seo *et al.*, 2006; Choi, Woo *et al.*, 2006), we report the crystal structure of the title compound, which has four unique molecules in the asymmetric unit (further marked as A, B, C and D) (Fig. 1).

The benzofuran unit is essentially planar, with a mean deviation of 0.018 (6) Å for A, 0.011 (6) Å for B, 0.010 (6) Å for C and 0.022 (6) Å for D, respectively, from the least-squares plane defined by the nine constituent atoms. The 4-bromophenyl rings are rotated out of the benzofuran planes, with dihedral angles of 20.8 (2), 6.7 (2), 23.5 (4) and 23.9 (4)° in the molecules A, B, C and D, respectively. The dihedral angles between the 3,4-methylenedioxyphenyl ring and the benzofuran plane are 13.5 (2), 7.1 (2), 18.6 (3) and 14.2 (3)° for the molecules A, B, C and D, respectively.

The molecular packing (Fig. 2) is stabilized by weak non-classical C–H···O hydrogen bonds; the first between the methylene H atom and the furan O atom, with a C12–H12A···O4, the second between the 4-bromophenyl H atom and the oxygen of the methylenedioxy group, with a C21–H21···O6<sup>i</sup>, the third between the methylene H atom and the furan O atom, with a C57–H57A···O10<sup>ii</sup>, the fourth between the 4-bromophenyl H atom and the oxygen of the methylenedioxy group, with a C65–H65···O11<sup>iii</sup>, the fifth between the methylene H atom and the furan O atom, with a C79–H79B···O7<sup>iv</sup>, respectively (Table 1).

The crystal packing (Fig. 3) exhibits aromatic  $\pi$ - $\pi$  interactions between the benzene/the furan rings and the 4-bromophenyl/the 3,4-methylenedioxyphenyl rings from two A molecules [Cg1…Cg2<sup>V</sup> = 3.92 (1) Å, Cg3…Cg4<sup>V</sup> = 3.79 (1) Å] (Cg1, Cg2, Cg3 and Cg4 are the centroids of the C2-C7 benzene ring, the C1/C2/C7/O1/C8 furan ring, the C16-C21 phenyl ring and the C9/C10/C11/C13/C14/C15 phenyl ring, respectively). The crystal packing (Fig. 3) also exhibits aromatic  $\pi$ - $\pi$  interactions between the benzene/the furan rings and the 4-bromophenyl/the 3,4-methylenedioxyphenyl rings from two B molecules [Cg5…Cg6<sup>Vi</sup> = 4.22 (1) Å, Cg7…Cg8<sup>Vi</sup> = 3.91 (1) Å] (Cg5, Cg6, Cg7 and Cg8 are the centroids of the C24-C29 benzene ring, the C23/C24/C29/O4/C30 furan ring, the C38-C43 phenyl ring and the C31/C32/C33/C34/C36/C37 phenyl ring, respectively). The crystal packing (Fig. 4) exhibits aromatic  $\pi$ - $\pi$  interactions between the benzene/the furan rings from two C molecules [Cg9…Cg10<sup>Vii</sup> = 3.77 (1) Å, Cg11…Cg12<sup>Vii</sup> = 3.77 (1) Å] (Cg9, Cg10, Cg11 and Cg12 are the centroids of the C46-C51 benzene ring, the C45/C46/C51/O7/C52 furan ring, the C60-C65 phenyl ring and the C53/C54/C55/C56/C58/C59 phenyl ring, respectively). The crystal packing (Fig. 4) exhibits aromatic  $\pi$ - $\pi$  interactions between the benzene/the benzene/the furan rings and the 4-bromophenyl/the 3,4-methylenedioxyphenyl ring, respectively). The crystal packing (Fig. 4) exhibits aromatic  $\pi$ - $\pi$  interactions between the benzene/the furan rings and the 4-bromophenyl/the 3,4-methylenedioxyphenyl rings from two C molecules [Cg9.cg10, Cg11 and Cg12 are the centroids of the C46-C51 benzene ring, the C45/C46/C51/O7/C52 furan ring, the C60-C65 phenyl ring and the C53/C54/C55/C56/C58/C59 phenyl ring, respectively). The crystal packing (Fig. 4) exhibits aromatic  $\pi$ - $\pi$  interactions between the benzene/the furan rings and the 4-bromophenyl/the 3,4-methylenedioxyphenyl rings from two D molecules [Cg13…Cg14<sup>Viii</sup> = 3.79 (1) Å, Cg15…Cg16<sup>Viii</sup> = 3.75 (1) Å] (Cg13, Cg14, Cg15 and

#### Experimental

Zinc chloride (409 mg, 3.0 mmol) was added to a stirred solution of 4'-bromo-[1,1'-biphenyl]-4-ol (747 mg, 3.0 mmol) and 2-chloro-2-methylsulfanyl-(3',4'-methylenedioxy)acetophenone (733 mg, 3.0 mmol) in dichloromethane (40 ml) at room temperature, and stirring was continued at the same temperature for 40 min. The reaction was quenched by the addition of water and the organic layer separated, dried over magnesium sulfate, filtered and concentrated in vacuum. The residue was purified by column chromatography (carbon tetrachloride) to afford the title compound as a colorless solid [yield 62%, m.p. 456-457 K;  $R_f = 0.35$  (carbon tetrachloride)]. Single crystals suitable for X-ray diffraction were prepared by evaporation of a solution of the title compound in tetrahydrofuran at room temperature. Spectroscopic analysis: <sup>1</sup>H NMR ( CDCl<sub>3</sub>, 400 MHz)  $\delta$  2.38 (s, 3H), 6.04 (s, 2H), 6.94 (d, J = 8.6 Hz, 1H), 7.49 (d, J = 1.96 Hz, 1H), 7.50-7.54 (m, 3H), 7.56-7.59 (m, 2H), 7.81 (d, J = 1.56 Hz, 1H), 7.84-7.87 (m, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  18.4, 101.4, 107.5, 107.9, 108.5, 111.4, 117.9, 121.2, 121.9, 124.1, 124.2, 129.1, 131.8, 131.9, 135.6, 140.4, 147.8, 148.4, 153.1, 156.1; EI-MS 438 [M<sup>+</sup>], 440 [M+2]

#### Refinement

All H atoms were geometrically positioned and refined using a riding model, with C-H = 0.95 Å for the aryl, 0.99 Å for the methylene, and 0.98 Å for the methyl H atoms. Uiso(H) =  $1.2U_{eq}(C)$  for the aryl and methylene H atoms, and  $1.5U_{eq}(C)$  for methyl H atoms.

#### **Figures**



Fig. 1. The molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are presented as a small cycles of arbitrary radius.



Fig. 2. C-H···O hydrogen bonds (dotted lines) in the title compound. [Symmetry codes: (i) - x + 2, -y + 1, -z; (ii) - x + 1, -y + 1, -z + 1; (iii) x, y + 1, z; (iv) - x, -y + 1, -z + 1.]



Fig. 3.  $\pi$ - $\pi$  interaction (dotted lines) of molecule A and B in the title compound. Cg denotes the ring centroids. [Symmetry codes: (v) - x + 1, - y + 1, - z; (vi) - x + 2, - y + 1, - z + 1.]



Fig. 4.  $\pi$ - $\pi$  interaction (dotted lines) of molecule C and D in the title compound. Cg denotes the ring centroids. [Symmetry codes: (vii) - x, -y + 2, -z + 1; (viii) - x + 1, -y, -z + 2.]

#### 5-(4-Bromophenyl)-2-(3,4-methylenedioxyphenyl)-3-methylsulfanyl-1- benzofuran

Crystal data	
C <sub>22</sub> H <sub>15</sub> BrO <sub>3</sub> S	Z = 8
$M_r = 439.31$	$F_{000} = 1776$
Triclinic, <i>P</i> T	$D_{\rm x} = 1.624 {\rm ~Mg~m}^{-3}$
Hall symbol: -P 1	Mo <i>K</i> $\alpha$ radiation, $\lambda = 0.71073$ Å
a = 12.3757 (8)  Å	Cell parameters from 3903 reflections
b = 16.067 (1)  Å	$\theta = 2.7 - 22.5^{\circ}$
c = 19.587 (1)  Å	$\mu = 2.43 \text{ mm}^{-1}$
$\alpha = 84.078 \ (1)^{\circ}$	T = 173  K
$\beta = 88.6573 \ (9)^{\circ}$	Block, colorless
$\gamma = 68.0552 \ (9)^{\circ}$	$0.24\times0.16\times0.12~mm$
$V = 3592.7 (4) \text{ Å}^3$	

#### Data collection

Bruker SMART CCD diffractometer	12562 independent reflections
Radiation source: fine-focus sealed tube	6573 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.094$
Detector resolution: 10.0 pixels mm <sup>-1</sup>	$\theta_{\text{max}} = 25.0^{\circ}$
T = 173  K	$\theta_{\min} = 1.1^{\circ}$
$\phi$ and $\omega$ scans	$h = -14 \rightarrow 14$
Absorption correction: multi-scan (SADABS; Sheldrick, 2000)	$k = -19 \rightarrow 19$
$T_{\min} = 0.594, \ T_{\max} = 0.760$	<i>l</i> = −23→23
26887 measured reflections	

#### Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.061$	H-atom parameters constrained
$wR(F^2) = 0.150$	$w = 1/[\sigma^2(F_o^2) + 15.1547P]$ where $P = (F_o^2 + 2F_c^2)/3$
S = 1.06	$(\Delta/\sigma)_{\rm max} < 0.001$

12562 reflections

973 parameters

 $\Delta \rho_{max} = 0.53 \text{ e Å}^{-3}$  $\Delta \rho_{min} = -0.73 \text{ e Å}^{-3}$ 

Primary atom site location: structure-invariant direct methods

#### Special details

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Extinction correction: none

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2 \text{sigma}(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{\rm iso}*/U_{\rm eq}$
Br1	0.50384 (8)	0.59231 (7)	-0.45348 (5)	0.0387 (3)
Br2	0.98757 (8)	0.61526 (7)	0.94332 (5)	0.0377 (3)
Br3	0.08943 (8)	0.90050 (7)	0.95449 (5)	0.0389 (3)
Br4	0.42898 (8)	0.09093 (7)	1.43447 (4)	0.0381 (3)
<b>S</b> 1	0.70263 (18)	0.55819 (14)	0.03947 (11)	0.0276 (5)
01	0.3585 (4)	0.6481 (4)	0.0621 (3)	0.0292 (14)
02	0.5119 (5)	0.6257 (4)	0.3701 (3)	0.0353 (15)
O3	0.6786 (5)	0.5523 (4)	0.3112 (3)	0.0407 (16)
C1	0.5504 (7)	0.5988 (5)	0.0350 (4)	0.0208 (19)
C2	0.4815 (7)	0.6134 (5)	-0.0271 (4)	0.023 (2)
C3	0.5089 (7)	0.6079 (5)	-0.0969 (4)	0.025 (2)
Н3	0.5876	0.5893	-0.1111	0.030*
C4	0.4193 (7)	0.6301 (5)	-0.1448 (4)	0.025 (2)
C5	0.3025 (7)	0.6563 (5)	-0.1211 (4)	0.029 (2)
Н5	0.2415	0.6683	-0.1538	0.035*
C6	0.2738 (7)	0.6649 (5)	-0.0528 (4)	0.029 (2)
H6	0.1954	0.6845	-0.0379	0.035*
C7	0.3664 (7)	0.6433 (6)	-0.0078 (4)	0.029 (2)
C8	0.4726 (7)	0.6201 (5)	0.0877 (4)	0.027 (2)
C9	0.4809 (7)	0.6233 (5)	0.1615 (4)	0.024 (2)
C10	0.5869 (7)	0.5820 (6)	0.1977 (4)	0.027 (2)
H10	0.6572	0.5514	0.1750	0.032*
C11	0.5854 (7)	0.5875 (6)	0.2657 (5)	0.030 (2)
C12	0.6349 (7)	0.5768 (7)	0.3773 (5)	0.043 (3)
H12A	0.6729	0.6148	0.3950	0.051*
H12B	0.6513	0.5221	0.4100	0.051*
C13	0.4852 (7)	0.6319 (5)	0.3001 (4)	0.0230 (19)
C14	0.3804 (8)	0.6730 (6)	0.2674 (4)	0.032 (2)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

1114	0.2120	0.7044	0.2012	0.020*
H14	0.3120	0.7044	0.2913	0.038*
C15	0.3779(7)	0.6670 (6)	0.1973 (4)	0.032 (2)
HIS	0.3056	0.6928	0.1729	0.039*
C16	0.4390 (7)	0.6245 (5)	-0.2201 (4)	0.0224 (19)
C17	0.3524 (7)	0.6734 (6)	-0.2691 (4)	0.028 (2)
H17	0.2794	0.7136	-0.2547	0.034*
C18	0.3712 (7)	0.6640 (6)	-0.3376 (5)	0.034 (2)
H18	0.3101	0.6952	-0.3701	0.040*
C19	0.4777 (7)	0.6098 (6)	-0.3593 (4)	0.031 (2)
C20	0.5679 (7)	0.5649 (5)	-0.3125 (4)	0.028 (2)
H20	0.6430	0.5297	-0.3278	0.034*
C21	0.5474 (7)	0.5717 (5)	-0.2431 (4)	0.028 (2)
H21	0.6086	0.5397	-0.2108	0.033*
C22	0.7251 (7)	0.6617 (6)	0.0456 (5)	0.040 (2)
H22A	0.8088	0.6491	0.0484	0.061*
H22B	0.6873	0.6886	0.0869	0.061*
H22C	0.6913	0.7038	0.0049	0.061*
S2	1.19277 (18)	0.55578 (15)	0.45491 (11)	0.0297 (5)
04	0.8489 (4)	0.6558 (4)	0.4222 (3)	0.0278 (14)
05	1.0111 (5)	0.6096 (4)	0.1168 (3)	0.0326 (15)
06	1.1794 (5)	0.5482 (4)	0.1839 (3)	0.0402 (16)
C23	1.0399 (7)	0.6017 (5)	0.4537 (4)	0.0224 (19)
C24	0.9693 (7)	0.6214 (5)	0.5151 (4)	0.0232 (19)
C25	0.9937 (7)	0.6146 (5)	0.5844 (4)	0.028 (2)
H25	1.0722	0.5932	0.6004	0.034*
C26	0.9024 (7)	0.6393 (5)	0.6310 (4)	0.0235 (19)
C27	0.7873 (7)	0.6691 (6)	0.6048 (4)	0.031 (2)
H27	0.7250	0.6841	0.6362	0.037*
C28	0.7610 (7)	0.6775 (6)	0.5356 (4)	0.032 (2)
H28	0.6830	0.6988	0.5188	0.038*
C29	0.8547 (7)	0.6531 (5)	0.4927 (4)	0.026 (2)
C30	0.9640 (7)	0.6234 (5)	0.4002 (4)	0.026 (2)
C31	0.9733 (7)	0.6229 (5)	0.3258 (4)	0.025 (2)
C32	0.8737 (7)	0.6596 (6)	0.2840 (5)	0.031 (2)
H32	0.8002	0.6862	0.3048	0.038*
C33	0.8777 (7)	0.6586 (6)	0.2127 (4)	0.028 (2)
H33	0.8089	0.6830	0.1849	0.034*
C34	0.9845 (7)	0.6211 (5)	0.1851 (4)	0.026 (2)
C35	1.1354 (7)	0.5715 (7)	0.1157 (5)	0.045 (3)
H35A	1.1615	0.5171	0.0906	0.054*
H35B	1.1654	0.6157	0.0918	0.054*
C36	1.0832 (7)	0.5839 (5)	0.2255 (4)	0.026 (2)
C37	1.0819 (7)	0.5830 (5)	0.2946 (4)	0.027 (2)
H37	1.1518	0.5563	0.3212	0.032*
C38	0.9236 (7)	0.6350 (5)	0.7062 (4)	0.0220 (19)
C39	1.0321 (7)	0.5840 (6)	0.7365 (4)	0.029 (2)
H39	1.0943	0.5521	0.7083	0.034*
C40	1.0520 (7)	0.5785 (5)	0.8059 (4)	0.029 (2)
H40	1.1269	0.5432	0.8251	0.035*

C41	0.9627 (7)	0.6243 (5)	0.8475 (4)	0.025 (2)
C42	0.8536 (7)	0.6781 (6)	0.8188 (4)	0.034 (2)
H42	0.7920	0.7109	0.8469	0.041*
C43	0.8366 (7)	0.6828 (6)	0.7496 (4)	0.032 (2)
H43	0.7625	0.7202	0.7303	0.038*
C44	1.2242 (8)	0.6572 (6)	0.4457 (6)	0.054 (3)
H44A	1.3088	0.6412	0.4456	0.082*
H44B	1.1907	0.6928	0.4841	0.082*
H44C	1.1903	0.6928	0.4024	0.082*
S3	0.26939 (18)	0.93222 (15)	0.45490 (11)	0.0268 (5)
07	-0.0074 (4)	0.8653 (3)	0.4399 (3)	0.0220 (13)
08	0.1301 (5)	0.8741 (4)	0.1293 (3)	0.0356 (15)
09	0.2265 (5)	0.9503 (4)	0.1814 (3)	0.0399 (16)
C45	0.1473 (6)	0.9008 (5)	0.4627 (4)	0.0184 (18)
C46	0.0944 (7)	0.8877 (5)	0.5277 (4)	0.0205 (19)
C47	0.1165 (7)	0.8896 (5)	0.5959 (4)	0.024 (2)
H47	0.1808	0.9036	0.6089	0.028*
C48	0.0439 (6)	0.8710 (5)	0.6462 (4)	0.0199 (18)
C49	-0.0528 (6)	0.8520 (5)	0.6248 (4)	0.0212 (19)
H49	-0.1038	0.8413	0.6585	0.025*
C50	-0.0743 (7)	0.8485 (5)	0.5572 (4)	0.024 (2)
H50	-0.1386	0.8348	0.5437	0.029*
C51	0.0002 (7)	0.8655 (5)	0.5095 (4)	0.025 (2)
C52	0.0862 (6)	0.8874 (5)	0.4125 (4)	0.0212 (19)
C53	0.0951 (6)	0.8823 (5)	0.3379 (4)	0.0206 (19)
C54	0.0372 (7)	0.8371 (5)	0.3053 (4)	0.025 (2)
Н54	-0.0083	0.8100	0.3320	0.030*
C55	0.0443 (7)	0.8308 (6)	0.2350 (4)	0.029 (2)
Н55	0.0041	0.8006	0.2131	0.034*
C56	0.1115 (7)	0.8699 (5)	0.1987 (4)	0.024 (2)
C57	0.2121 (8)	0.9185 (8)	0.1186 (4)	0.050 (3)
H57A	0.2878	0.8757	0.1040	0.060*
H57B	0.1825	0.9696	0.0823	0.060*
C58	0.1688 (7)	0.9152 (5)	0.2306 (4)	0.0226 (19)
C59	0.1631 (7)	0.9225 (5)	0.2981 (4)	0.026 (2)
Н59	0.2035	0.9537	0.3187	0.031*
C60	0.0611 (6)	0.8750 (5)	0.7200 (4)	0.0212 (19)
C61	0.0161 (7)	0.8278 (5)	0.7701 (4)	0.026 (2)
H61	-0.0217	0.7905	0.7562	0.032*
C62	0.0257 (8)	0.8349 (6)	0.8384 (4)	0.035 (2)
H62	-0.0073	0.8039	0.8713	0.042*
C63	0.0827 (7)	0.8866 (6)	0.8599 (4)	0.027 (2)
C64	0.1315 (7)	0.9312 (6)	0.8122 (4)	0.028 (2)
H64	0.1726	0.9657	0.8269	0.034*
C65	0.1204 (7)	0.9255 (5)	0.7438 (4)	0.026 (2)
H65	0.1540	0.9567	0.7115	0.031*
C66	0.3844 (7)	0.8217 (6)	0.4697 (5)	0.045(3)
H66A	0.4603	0.8278	0.4669	0.068*
H66B	0.3790	0.7838	0.4347	0.068*

H66C	0.3761	0.7937	0.5153	0.068*
S4	0.23241 (18)	0.06678 (15)	0.94641 (11)	0.0272 (5)
O10	0.5046 (4)	0.1399 (4)	0.9101 (3)	0.0260 (13)
011	0.2639 (5)	0.0615 (4)	0.6721 (3)	0.0390 (16)
012	0.3684 (5)	0.1314 (4)	0.6048 (3)	0.0318 (15)
C67	0.3528 (6)	0.1008 (5)	0.9437 (4)	0.0207 (19)
C68	0.4092 (7)	0.1112 (5)	1.0048 (4)	0.024 (2)
C69	0.3869 (7)	0.1089 (5)	1.0745 (4)	0.0231 (19)
H69	0.3228	0.0949	1.0920	0.028*
C70	0.4605 (7)	0.1275 (5)	1.1188 (4)	0.0232 (19)
C71	0.5530 (7)	0.1510 (5)	1.0910 (4)	0.025 (2)
H71	0.6023	0.1638	1.1212	0.030*
C72	0.5741 (6)	0.1559 (5)	1.0216 (4)	0.027 (2)
H72	0.6364	0.1718	1.0034	0.032*
C73	0.4992 (6)	0.1362 (5)	0.9795 (4)	0.0209 (19)
C74	0.4106 (7)	0.1194 (5)	0.8889 (4)	0.024 (2)
C75	0.4005 (7)	0.1234 (5)	0.8139 (4)	0.0231 (19)
C76	0.4619 (7)	0.1663 (6)	0.7723 (4)	0.031 (2)
H76	0.5087	0.1919	0.7934	0.037*
C77	0.4563 (7)	0.1724 (5)	0.7009 (4)	0.029 (2)
H77	0.4998	0.2001	0.6728	0.035*
C78	0.3859 (7)	0.1370 (5)	0.6735 (4)	0.025 (2)
C79	0.2758 (8)	0.0978 (7)	0.6031 (4)	0.041 (2)
H79A	0.2950	0.0502	0.5715	0.049*
H79B	0.2022	0.1471	0.5871	0.049*
C80	0.3247 (7)	0.0941 (5)	0.7139 (4)	0.026 (2)
C81	0.3293 (7)	0.0861 (5)	0.7832 (4)	0.028 (2)
H81	0.2864	0.0568	0.8101	0.033*
C82	0.4455 (6)	0.1239 (5)	1.1945 (4)	0.0208 (19)
C83	0.4896 (7)	0.1713 (6)	1.2348 (4)	0.030 (2)
H83	0.5261	0.2093	1.2129	0.036*
C84	0.4817 (7)	0.1646 (6)	1.3051 (4)	0.033 (2)
H84	0.5108	0.1983	1.3314	0.039*
C85	0.4309 (7)	0.1082 (6)	1.3366 (4)	0.027 (2)
C86	0.3821 (7)	0.0630 (5)	1.2990 (4)	0.028 (2)
H86	0.3441	0.0265	1.3214	0.034*
C87	0.3887 (7)	0.0711 (5)	1.2290 (4)	0.026 (2)
H87	0.3542	0.0404	1.2032	0.031*
C88	0.1155 (7)	0.1741 (6)	0.9358 (5)	0.048 (3)
H88A	0.0411	0.1655	0.9363	0.072*
H88B	0.1231	0.2065	0.8919	0.072*
H88C	0.1180	0.2092	0.9734	0.072*
$(1, 1, 1, \dots, n, n) \in \mathbb{R}^2$				
Atomic displacement parameters $(A^-)$				

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	U <sup>23</sup>
Br1	0.0384 (6)	0.0440 (6)	0.0303 (5)	-0.0108 (5)	0.0001 (4)	-0.0061 (5)
Br2	0.0373 (6)	0.0498 (6)	0.0261 (5)	-0.0166 (5)	0.0002 (4)	-0.0027 (5)

Br3	0.0489 (6)	0.0550 (7)	0.0248 (5)	-0.0313 (5)	0.0044 (5)	-0.0127 (5)
Br4	0.0449 (6)	0.0502 (6)	0.0241 (5)	-0.0239 (5)	-0.0008 (4)	-0.0008 (4)
S1	0.0227 (12)	0.0264 (13)	0.0307 (13)	-0.0048 (10)	0.0014 (10)	-0.0068 (10)
01	0.023 (3)	0.042 (4)	0.028 (3)	-0.018 (3)	0.003 (3)	-0.006 (3)
O2	0.037 (4)	0.046 (4)	0.025 (3)	-0.017 (3)	0.002 (3)	-0.007 (3)
O3	0.034 (4)	0.061 (5)	0.028 (4)	-0.018 (3)	0.000 (3)	-0.003 (3)
C1	0.024 (5)	0.016 (4)	0.026 (5)	-0.010 (4)	-0.001 (4)	-0.004 (4)
C2	0.030 (5)	0.014 (5)	0.026 (5)	-0.007 (4)	0.004 (4)	-0.007 (4)
C3	0.023 (5)	0.015 (5)	0.032 (5)	-0.001 (4)	0.006 (4)	-0.006 (4)
C4	0.026 (5)	0.020 (5)	0.030 (5)	-0.009 (4)	0.003 (4)	-0.004 (4)
C5	0.024 (5)	0.030 (5)	0.036 (6)	-0.012 (4)	0.000 (4)	-0.008 (4)
C6	0.017 (5)	0.028 (5)	0.045 (6)	-0.010 (4)	0.000 (4)	-0.008 (4)
C7	0.015 (5)	0.035 (5)	0.039 (6)	-0.010 (4)	0.009 (4)	-0.016 (4)
C8	0.019 (5)	0.027 (5)	0.037 (6)	-0.010 (4)	-0.004 (4)	-0.009 (4)
C9	0.025 (5)	0.021 (5)	0.027 (5)	-0.008 (4)	0.000 (4)	-0.003 (4)
C10	0.021 (5)	0.033 (5)	0.030 (5)	-0.013 (4)	0.010 (4)	-0.011 (4)
C11	0.022 (5)	0.026 (5)	0.043 (6)	-0.012 (4)	-0.005 (4)	-0.002 (4)
C12	0.028 (5)	0.067 (7)	0.034 (6)	-0.021 (5)	0.001 (5)	-0.004 (5)
C13	0.026 (5)	0.022 (5)	0.023 (5)	-0.010 (4)	0.001 (4)	-0.004 (4)
C14	0.035 (5)	0.038 (6)	0.026 (5)	-0.016 (5)	0.008 (4)	-0.009 (4)
C15	0.029 (5)	0.028 (5)	0.036 (6)	-0.007 (4)	0.000 (4)	-0.005 (4)
C16	0.023 (5)	0.019 (5)	0.030 (5)	-0.013 (4)	-0.002 (4)	-0.005 (4)
C17	0.024 (5)	0.031 (5)	0.028 (5)	-0.006 (4)	-0.005 (4)	-0.007 (4)
C18	0.030 (5)	0.032 (6)	0.034 (6)	-0.006 (4)	-0.001 (4)	-0.007 (4)
C19	0.031 (5)	0.027 (5)	0.034 (5)	-0.010 (4)	0.003 (4)	-0.005 (4)
C20	0.023 (5)	0.027 (5)	0.036 (5)	-0.011 (4)	0.008 (4)	-0.006 (4)
C21	0.024 (5)	0.029 (5)	0.026 (5)	-0.005 (4)	-0.007 (4)	0.002 (4)
C22	0.029 (5)	0.040 (6)	0.065 (7)	-0.025 (5)	0.008 (5)	-0.016 (5)
S2	0.0232 (12)	0.0299 (13)	0.0319 (13)	-0.0052 (10)	-0.0029 (10)	-0.0023 (10)
O4	0.019 (3)	0.039 (4)	0.030 (3)	-0.016 (3)	0.005 (3)	-0.007 (3)
O5	0.025 (3)	0.039 (4)	0.026 (3)	-0.004 (3)	0.001 (3)	-0.001 (3)
O6	0.025 (3)	0.056 (4)	0.027 (4)	0.000 (3)	0.000 (3)	-0.010 (3)
C23	0.022 (5)	0.021 (5)	0.026 (5)	-0.009 (4)	-0.005 (4)	-0.004 (4)
C24	0.028 (5)	0.021 (5)	0.024 (5)	-0.011 (4)	0.001 (4)	-0.007 (4)
C25	0.019 (5)	0.025 (5)	0.045 (6)	-0.013 (4)	-0.002 (4)	-0.001 (4)
C26	0.029 (5)	0.010 (4)	0.031 (5)	-0.007 (4)	0.003 (4)	-0.003 (4)
C27	0.026 (5)	0.045 (6)	0.029 (5)	-0.020 (4)	0.003 (4)	-0.009 (4)
C28	0.018 (5)	0.048 (6)	0.035 (6)	-0.020 (4)	0.006 (4)	-0.006 (5)
C29	0.031 (5)	0.027 (5)	0.024 (5)	-0.017 (4)	-0.001 (4)	-0.003 (4)
C30	0.028 (5)	0.021 (5)	0.031 (5)	-0.010 (4)	0.008 (4)	-0.008 (4)
C31	0.027 (5)	0.019 (5)	0.029 (5)	-0.009 (4)	0.002 (4)	0.000 (4)
C32	0.018 (5)	0.030 (5)	0.043 (6)	-0.006 (4)	0.001 (4)	-0.001 (4)
C33	0.016 (5)	0.040 (6)	0.024 (5)	-0.005 (4)	-0.007 (4)	0.001 (4)
C34	0.030 (5)	0.030 (5)	0.016 (5)	-0.009 (4)	-0.001 (4)	-0.008 (4)
C35	0.028 (5)	0.069 (7)	0.035 (6)	-0.014 (5)	0.008 (5)	-0.006 (5)
C36	0.022 (5)	0.026 (5)	0.026 (5)	-0.005 (4)	0.001 (4)	0.001 (4)
C37	0.018 (5)	0.028 (5)	0.031 (5)	-0.004 (4)	-0.005 (4)	-0.004 (4)
C38	0.029 (5)	0.020 (5)	0.021 (5)	-0.015 (4)	-0.005 (4)	0.004 (4)
C39	0.038 (5)	0.028 (5)	0.024 (5)	-0.014 (4)	0.008 (4)	-0.008 (4)

C40	0.024 (5)	0.026 (5)	0.034 (5)	-0.007 (4)	-0.004 (4)	0.004 (4)
C41	0.028 (5)	0.024 (5)	0.029 (5)	-0.014 (4)	0.007 (4)	-0.005 (4)
C42	0.028 (5)	0.037 (6)	0.032 (6)	-0.007 (4)	0.002 (4)	-0.001 (4)
C43	0.022 (5)	0.039 (6)	0.029 (5)	-0.007 (4)	-0.002 (4)	0.004 (4)
C44	0.030 (6)	0.042 (7)	0.092 (9)	-0.017 (5)	-0.017 (6)	0.011 (6)
S3	0.0297 (12)	0.0319 (13)	0.0274 (12)	-0.0214 (11)	0.0014 (10)	-0.0031 (10)
07	0.017 (3)	0.031 (3)	0.020 (3)	-0.013 (3)	0.000 (2)	-0.001 (3)
08	0.040 (4)	0.056 (4)	0.018 (3)	-0.024 (3)	0.010 (3)	-0.012 (3)
09	0.048 (4)	0.065 (5)	0.023 (3)	-0.039 (4)	0.006 (3)	-0.008 (3)
C45	0.022 (4)	0.017 (4)	0.017 (4)	-0.009 (4)	0.001 (4)	-0.002 (3)
C46	0.021 (5)	0.018 (5)	0.023 (5)	-0.008 (4)	-0.004 (4)	-0.002 (4)
C47	0.017 (4)	0.023 (5)	0.034 (5)	-0.011 (4)	-0.001 (4)	-0.008 (4)
C48	0.019 (4)	0.013 (4)	0.028 (5)	-0.005 (4)	-0.003 (4)	0.000 (4)
C49	0.016 (4)	0.022 (5)	0.030 (5)	-0.013 (4)	0.005 (4)	0.001 (4)
C50	0.021 (5)	0.032 (5)	0.025 (5)	-0.016 (4)	0.002 (4)	-0.005 (4)
C51	0.021 (5)	0.028 (5)	0.021 (5)	-0.005 (4)	0.003 (4)	-0.004 (4)
C52	0.016 (4)	0.017 (5)	0.028 (5)	-0.005 (4)	0.011 (4)	0.005 (4)
C53	0.020 (4)	0.021 (5)	0.023 (5)	-0.009 (4)	0.004 (4)	-0.005 (4)
C54	0.023 (5)	0.030 (5)	0.028 (5)	-0.016 (4)	0.001 (4)	-0.005 (4)
C55	0.031 (5)	0.031 (5)	0.029 (5)	-0.016 (4)	-0.004 (4)	-0.006 (4)
C56	0.031 (5)	0.021 (5)	0.019 (5)	-0.008 (4)	-0.001 (4)	-0.002 (4)
C57	0.045 (6)	0.097 (9)	0.022 (5)	-0.040 (6)	0.016 (5)	-0.019 (5)
C58	0.023 (5)	0.027 (5)	0.020 (5)	-0.012 (4)	0.000 (4)	-0.001 (4)
C59	0.021 (5)	0.031 (5)	0.027 (5)	-0.011 (4)	-0.004 (4)	-0.004 (4)
C60	0.019 (4)	0.015 (5)	0.029 (5)	-0.007 (4)	0.006 (4)	0.001 (4)
C61	0.027 (5)	0.034 (5)	0.028 (5)	-0.022 (4)	-0.005 (4)	-0.004 (4)
C62	0.049 (6)	0.042 (6)	0.025 (5)	-0.031 (5)	-0.002 (5)	0.005 (4)
C63	0.034 (5)	0.030 (5)	0.020 (5)	-0.015 (4)	-0.001 (4)	-0.007 (4)
C64	0.028 (5)	0.036 (6)	0.028 (5)	-0.020 (4)	-0.005 (4)	-0.004 (4)
C65	0.027 (5)	0.028 (5)	0.028 (5)	-0.016 (4)	0.006 (4)	-0.005 (4)
C66	0.017 (5)	0.044 (6)	0.076 (8)	-0.012 (5)	0.008 (5)	-0.013 (5)
S4	0.0304 (12)	0.0308 (13)	0.0290 (12)	-0.0212 (11)	0.0007 (10)	-0.0028 (10)
O10	0.028 (3)	0.033 (4)	0.025 (3)	-0.019 (3)	0.004 (3)	-0.009 (3)
011	0.049 (4)	0.061 (5)	0.022 (3)	-0.038 (4)	-0.003 (3)	-0.002 (3)
012	0.034 (4)	0.051 (4)	0.018 (3)	-0.026 (3)	-0.001 (3)	-0.003 (3)
C67	0.020 (4)	0.018 (5)	0.021 (5)	-0.004 (4)	-0.004 (4)	-0.003 (4)
C68	0.021 (5)	0.016 (5)	0.034 (5)	-0.006 (4)	0.003 (4)	0.001 (4)
C69	0.021 (5)	0.016 (5)	0.031 (5)	-0.006 (4)	0.004 (4)	0.000 (4)
C70	0.026 (5)	0.016 (5)	0.026 (5)	-0.006 (4)	-0.006 (4)	-0.001 (4)
C71	0.026 (5)	0.025 (5)	0.031 (5)	-0.014 (4)	0.002 (4)	-0.013 (4)
C72	0.012 (4)	0.034 (5)	0.037 (5)	-0.010 (4)	0.005 (4)	-0.004 (4)
C73	0.016 (4)	0.024 (5)	0.023 (5)	-0.007 (4)	-0.004 (4)	-0.001 (4)
C74	0.017 (4)	0.028 (5)	0.032 (5)	-0.012 (4)	-0.003 (4)	-0.005 (4)
C75	0.022 (5)	0.022 (5)	0.026 (5)	-0.010 (4)	0.000 (4)	-0.004 (4)
C/6	0.033 (5)	0.030 (5)	0.034 (5)	-0.017 (4)	0.007 (4)	-0.010 (4)
C77	0.040 (5)	0.032 (5)	0.027 (5)	-0.026 (5)	0.011 (4)	-0.006 (4)
C/8	0.024 (5)	0.030 (5)	0.021 (5)	-0.010 (4)	0.005 (4)	-0.007(4)
C/9	0.044 (6)	0.068 (7)	0.027 (5)	-0.036 (6)	0.002 (5)	-0.010(5)
080	0.024 (5)	0.021 (5)	0.035 (5)	-0.011 (4)	0.001 (4)	-0.009 (4)

C81	0.022 (5)	0.030 (5)	0.033 (5)		-0.011 (4)	-0.002 (4)	-0.006 (4	)
C82	0.018 (4)	0.016 (5)	0.027 (5)		-0.004 (4)	-0.004 (4)	-0.004 (4	)
C83	0.040 (5)	0.028 (5)	0.025 (5)		-0.016 (4)	0.004 (4)	-0.005 (4	)
C84	0.046 (6)	0.032 (6)	0.032 (5)		-0.029 (5)	0.001 (4)	-0.003 (4	)
C85	0.033 (5)	0.030 (5)	0.024 (5)		-0.019 (4)	0.002 (4)	-0.001 (4	)
C86	0.034 (5)	0.026 (5)	0.028 (5)		-0.017 (4)	0.001 (4)	0.003 (4)	
C87	0.033 (5)	0.019 (5)	0.026 (5)		-0.009 (4)	-0.002 (4)	-0.006 (4	)
C88	0.023 (5)	0.040 (6)	0.078 (8)		-0.011 (5)	-0.016 (5)	0.008 (6)	
Gaometric naran	natars (Å °)							
Geometric purun	neiers (A, )							
Br1—C19		1.897 (9)		C44—H4	14B		0.9800	
Br2—C41		1.890 (8)		C44—H4	14C		0.9800	
Br3—C63		1.895 (8)		S3—C45			1.763 (8)	
Br4—C85		1.909 (8)		S3—C66			1.813 (9)	
S1—C1		1.749 (8)		07—C51	l		1.370 (9)	
S1—C22		1.802 (8)		O7—C52	2		1.412 (8)	
O1—C7		1.378 (9)		08—C56	6		1.373 (9)	
O1—C8		1.399 (9)		08—C57	7		1.443 (10)	
O2—C13		1.403 (9)		O9—C58	3		1.383 (9)	
O2—C12		1.431 (9)		O9—C57	7		1.419 (10)	
O3—C11		1.379 (9)		C45—C5	52		1.337 (10)	
O3—C12		1.428 (10)		C45—C4	16		1.450 (10)	
C1—C8		1.376 (10)		C46—C4	17		1.376 (10)	
C1—C2		1.450 (10)		C46—C5	51		1.403 (10)	
C2—C7		1.379 (10)		C47—C4	18		1.401 (10)	
C2—C3		1.405 (11)		C47—H4	17		0.9500	
C3—C4		1.386 (11)		C48—C4	19		1.423 (10)	
С3—Н3		0.9500		C48—C6	50		1.477 (11)	
C4—C5		1.427 (10)		C49—C5	50		1.367 (10)	
C4—C16		1.496 (11)		C49—H4	19		0.9500	
C5—C6		1.383 (11)		C50—C5	51		1.377 (10)	
С5—Н5		0.9500		С50—Н3	50		0.9500	
С6—С7		1.378 (11)		C52—C5	53		1.471 (10)	
С6—Н6		0.9500		C53—C5	54		1.396 (10)	
С8—С9		1.458 (11)		C53—C5	59		1.417 (10)	
C9—C10		1.400 (11)		C54—C5	55		1.389 (11)	
C9—C15		1.414 (11)		С54—Н5	54		0.9500	
C10-C11		1.343 (11)		C55—C5	56		1.363 (11)	
C10—H10		0.9500		С55—Н5	55		0.9500	
C11—C13		1.379 (11)		C56—C5	58		1.386 (10)	
C12—H12A		0.9900		С57—Н5	57A		0.9900	
C12—H12B		0.9900		С57—Н	57B		0.9900	
C13—C14		1.357 (11)		C58—C5	59		1.337 (10)	
C14—C15		1.388 (11)		С59—Н5	59		0.9500	
C14—H14		0.9500		C60—C6	55		1.398 (10)	
C15—H15		0.9500		C60—C6	51		1.409 (10)	
C16—C21		1.387 (10)		C61—C6	52		1.366 (11)	
C16—C17		1.396 (10)		С61—Не	51		0.9500	

C17—C18	1.371 (11)	C62—C63	1.374 (11)
C17—H17	0.9500	С62—Н62	0.9500
C18—C19	1.366 (11)	C63—C64	1.383 (11)
C18—H18	0.9500	C64—C65	1.365 (11)
C19—C20	1.384 (11)	C64—H64	0.9500
C20—C21	1.385 (11)	С65—Н65	0.9500
С20—Н20	0.9500	С66—Н66А	0.9800
C21—H21	0.9500	С66—Н66В	0.9800
C22—H22A	0.9800	С66—Н66С	0.9800
C22—H22B	0.9800	S4—C67	1.766 (8)
С22—Н22С	0.9800	S4—C88	1.785 (9)
S2—C23	1.755 (8)	O10—C73	1.356 (9)
S2—C44	1.803 (9)	O10—C74	1.403 (8)
O4—C29	1.379 (9)	O11—C80	1.386 (9)
O4—C30	1.396 (9)	O11—C79	1.443 (9)
O5—C34	1.385 (9)	O12—C78	1.385 (9)
O5—C35	1.428 (9)	O12—C79	1.439 (9)
O6—C36	1.396 (9)	C67—C74	1.346 (10)
O6—C35	1.414 (10)	C67—C68	1.455 (11)
C23—C30	1.351 (11)	C68—C69	1.387 (11)
C23—C24	1.461 (10)	C68—C73	1.383 (10)
C24—C25	1.382 (11)	C69—C70	1.402 (10)
C24—C29	1.381 (11)	С69—Н69	0.9500
C25—C26	1.398 (10)	C70—C71	1.414 (10)
C25—H25	0.9500	C70—C82	1.489 (11)
C26—C27	1.413 (11)	C71—C72	1.377 (11)
C26—C38	1.493 (11)	C71—H71	0.9500
C27—C28	1.382 (11)	C72—C73	1.396 (10)
C27—H27	0.9500	С72—Н72	0.9500
C28—C29	1.373 (10)	C74—C75	1.470 (11)
C28—H28	0.9500	C75—C76	1.401 (11)
C30—C31	1.460 (11)	C75—C81	1.410 (10)
C31—C32	1.394 (11)	C76—C77	1.394 (11)
C31—C37	1.410 (10)	С76—Н76	0.9500
C32—C33	1.398 (11)	C77—C78	1.351 (10)
С32—Н32	0.9500	С77—Н77	0.9500
C33—C34	1.357 (10)	C78—C80	1.387 (11)
С33—Н33	0.9500	С79—Н79А	0.9900
C34—C36	1.370 (10)	С79—Н79В	0.9900
С35—Н35А	0.9900	C80—C81	1.351 (11)
С35—Н35В	0.9900	C81—H81	0.9500
C36—C37	1.351 (11)	C82—C87	1.405 (10)
С37—Н37	0.9500	C82—C83	1.398 (10)
C38—C43	1.395 (11)	C83—C84	1.375 (11)
C38—C39	1.394 (11)	С83—Н83	0.9500
C39—C40	1.376 (11)	C84—C85	1.374 (11)
С39—Н39	0.9500	C84—H84	0.9500
C40—C41	1.379 (11)	C85—C86	1.377 (10)
C40—H40	0.9500	C86—C87	1.366 (11)

C41—C42	1.397 (11)	С86—Н86	0.9500
C42—C43	1.365 (11)	С87—Н87	0.9500
C42—H42	0.9500	C88—H88A	0.9800
С43—Н43	0.9500	C88—H88B	0.9800
C44—H44A	0.9800	C88—H88C	0.9800
C1—S1—C22	100.3 (4)	C45—S3—C66	99.4 (4)
C7—O1—C8	106.8 (6)	C51—O7—C52	105.6 (6)
C13—O2—C12	106.1 (6)	C56—O8—C57	105.2 (6)
C11—O3—C12	107.0 (6)	C58—O9—C57	106.5 (6)
C8—C1—C2	106.3 (7)	C52—C45—C46	108.5 (7)
C8—C1—S1	128.4 (6)	C52—C45—S3	127.8 (6)
C2-C1-S1	125.3 (6)	C46—C45—S3	123.6 (6)
C7—C2—C3	119.6 (8)	C47—C46—C51	119.3 (7)
C7-C2-C1	106.6 (7)	C47—C46—C45	136.6 (7)
$C_{3} - C_{2} - C_{1}$	133.7 (8)	C51—C46—C45	104.0 (7)
C4-C3-C2	1190(7)	C46-C47-C48	1198(7)
C4—C3—H3	120.5	C46—C47—H47	120.1
C2-C3-H3	120.5	C48—C47—H47	120.1
$C_{3}$ $C_{4}$ $C_{5}$	118.6 (7)	C47 - C48 - C49	120.1 118.6(7)
$C_{3}$ $C_{4}$ $C_{16}$	123 3 (7)	C47 - C48 - C60	1222(7)
$C_{5} - C_{4} - C_{16}$	123.3(7) 1181(7)	C49 - C48 - C60	122.2(7) 1191(7)
$C_{6}$	123.0 (8)	$C_{50}$ $C_{49}$ $C_{48}$	122.0(7)
C6_C5_H5	118.5	$C_{50}$ $C_{49}$ $H_{49}$	119.0
C4-C5-H5	118.5	$C_{48}$ $C_{49}$ $H_{49}$	119.0
C7C6C5	115.6 (8)	$C_{40} = C_{50} = C_{51}$	117.0
C7-C6-H6	122.2	C49 - C50 - H50	121.2
C5-C6-H6	122.2	C51-C50-H50	121.2
$C_{6}$	122.2 125.7(7)	07-051-050	121.2 126.1(7)
$C_{6}^{-}$ $C_{7}^{-}$ $C_{2}^{-}$	123.7(7) 124.0(8)	07 - 051 - 046	120.1(7)
01 - 07 - 02	124.0(8) 1103(7)	57 - 51 - 546	111.4(7) 122.6(7)
$C_1 - C_2 = C_2$	110.5(7)	$C_{45} = C_{52} = C_{7}$	122.0(7) 110.4(7)
$C_1 - C_8 - C_9$	135.6 (7)	C45 - C52 - C53	137.6(7)
01 - (8 - 6)	1143(7)	07 - 052 - 053	137.0(7) 111.8(7)
$C_{10} - C_{9} - C_{15}$	114.5(7)	57 - 53 - 59	1100(7)
$C_{10} = C_{9} = C_{13}$	119.0(7) 121.7(7)	$C_{54} = C_{53} = C_{54}$	119.0(7) 120.5(7)
$C_{10} = C_{9} = C_{8}$	121.7(7) 1187(7)	$C_{54} = C_{55} = C_{52}$	120.5(7)
$C_{13} = C_{3} = C_{3}$	110.7(7) 117.2(8)	$C_{55} = C_{52} = C_{52}$	120.3(7) 121.8(8)
$C_{11} = C_{10} = C_{3}$	117.3 (8)	C55 C54 H54	121.0 (0)
$C_{11}$ $C_{10}$ $H_{10}$	121.5	C53_C54_H54	119.1
$C_{10} = C_{10} = C_{10}$	121.5	C55—C54—I154	117.2 (8)
$C_{10} = C_{11} = C_{13}$	122.9(8)	C56 C55 H55	121 4
$C_{10} = C_{11} = 03$	127.3(8) 100 8(7)	C54 C55 H55	121.4
$0^{2}$ $0^{12}$ $0^{2}$	107.8(7)	C55 C56 O8	121.4 128.0(7)
03 - 012 - 02	107.8(7)	$C_{55} = C_{56} = C_{58}$	120.0(7)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	110.2	$C_{3}$ $C_{50}$ $C_$	121.3(7)
$O_2 = C_{12} = M_{12} P_{12}$	110.2	09  C57  09	110.0(7) 108.1(6)
03 - 012 -	110.2	09 - 03 - 00	100.1 (0)
$U_2 \rightarrow U_1 2 \rightarrow U_1 2 D$	110.2	07 - 03 - 03 - 03 - 03 - 03 - 03 - 03 -	110.1
$\Pi 12A - C 12 - \Pi 12B$	100.0	00 - 057 - H57P	110.1
014-013-011	122.0 (8)	Uy—U3/—H3/B	110.1

C14—C13—O2	128.7 (7)	O8—C57—H57B	110.1
C11—C13—O2	109.3 (7)	Н57А—С57—Н57В	108.4
C13—C14—C15	116.9 (8)	C59—C58—O9	128.5 (7)
C13—C14—H14	121.6	C59—C58—C56	122.7 (8)
C15-C14-H14	121.6	O9—C58—C56	108.9 (7)
C14—C15—C9	121.3 (8)	C58—C59—C53	117.9 (8)
C14—C15—H15	119.4	С58—С59—Н59	121.0
С9—С15—Н15	119.4	С53—С59—Н59	121.0
C21—C16—C17	118.1 (8)	C65—C60—C61	116.7 (7)
C21—C16—C4	119.8 (7)	C65—C60—C48	122.7 (7)
C17—C16—C4	122.1 (7)	C61—C60—C48	120.6 (7)
C18—C17—C16	120.8 (8)	C62—C61—C60	121.2 (7)
C18—C17—H17	119.6	C62—C61—H61	119.4
С16—С17—Н17	119.6	С60—С61—Н61	119.4
C17—C18—C19	120.2 (8)	C61—C62—C63	120.5 (8)
C17—C18—H18	119.9	С61—С62—Н62	119.8
C19-C18-H18	119.9	С63—С62—Н62	119.8
C18-C19-C20	120.4 (8)	C62—C63—C64	119.8 (7)
C18—C19—Br1	120.8 (7)	C62—C63—Br3	120.2 (6)
C20-C19-Br1	118.7 (6)	C64—C63—Br3	120.0 (6)
C21—C20—C19	119.3 (8)	C65—C64—C63	119.9 (8)
C21—C20—H20	120.4	С65—С64—Н64	120.1
С19—С20—Н20	120.4	С63—С64—Н64	120.1
C20-C21-C16	121.0 (8)	C64—C65—C60	121.9 (8)
C20—C21—H21	119.5	С64—С65—Н65	119.1
C16—C21—H21	119.5	С60—С65—Н65	119.1
S1—C22—H22A	109.5	S3—C66—H66A	109.5
S1—C22—H22B	109.5	S3—C66—H66B	109.5
H22A—C22—H22B	109.5	Н66А—С66—Н66В	109.5
S1—C22—H22C	109.5	S3—C66—H66C	109.5
H22A—C22—H22C	109.5	Н66А—С66—Н66С	109.5
H22B—C22—H22C	109.5	Н66В—С66—Н66С	109.5
C23—S2—C44	100.7 (4)	C67—S4—C88	100.3 (4)
C29—O4—C30	106.1 (6)	C73—O10—C74	105.6 (6)
C34—O5—C35	104.7 (6)	C80—O11—C79	105.7 (6)
C36—O6—C35	105.6 (6)	C78—O12—C79	105.6 (6)
C30—C23—C24	106.2 (7)	C74—C67—C68	107.5 (7)
C30—C23—S2	129.9 (6)	C74—C67—S4	129.2 (6)
C24—C23—S2	123.9 (6)	C68—C67—S4	123.3 (6)
C25—C24—C29	119.4 (8)	C69—C68—C73	120.0 (8)
C25—C24—C23	134.7 (8)	C69—C68—C67	135.7 (8)
C29—C24—C23	106.0 (7)	C73—C68—C67	104.1 (7)
C24—C25—C26	119.8 (8)	C68—C69—C70	119.0 (7)
C24—C25—H25	120.1	С68—С69—Н69	120.5
C26—C25—H25	120.1	С70—С69—Н69	120.5
C25—C26—C27	117.9 (8)	C69—C70—C71	119.2 (7)
C25—C26—C38	122.1 (7)	C69—C70—C82	122.7 (7)
C27—C26—C38	120.0 (7)	C71—C70—C82	118.1 (7)
C28—C27—C26	123.2 (8)	C72—C71—C70	122.4 (7)

С28—С27—Н27	118.4	С72—С71—Н71	118.8
С26—С27—Н27	118.4	С70—С71—Н71	118.8
C27—C28—C29	115.9 (8)	C71—C72—C73	116.5 (7)
C27—C28—H28	122.1	С71—С72—Н72	121.8
C29—C28—H28	122.1	С73—С72—Н72	121.8
O4—C29—C28	125.7 (7)	O10—C73—C68	112.3 (7)
O4—C29—C24	110.5 (7)	O10—C73—C72	124.8 (7)
C28—C29—C24	123.8 (8)	C68—C73—C72	123.0 (8)
C23—C30—O4	111.3 (7)	C67—C74—O10	110.4 (7)
C23—C30—C31	135.6 (8)	C67—C74—C75	136.4 (7)
O4—C30—C31	113.1 (7)	O10—C74—C75	113.2 (7)
C32—C31—C37	118.3 (8)	C76—C75—C81	119.6 (8)
C32—C31—C30	120.3 (7)	C76—C75—C74	118.9 (7)
C37—C31—C30	121.3 (7)	C81—C75—C74	121.5 (7)
C31—C32—C33	122.5 (8)	C77—C76—C75	122.0 (8)
С31—С32—Н32	118.8	С77—С76—Н76	119.0
С33—С32—Н32	118.8	С75—С76—Н76	119.0
C34—C33—C32	116.8 (8)	C78—C77—C76	116.5 (8)
С34—С33—Н33	121.6	С78—С77—Н77	121.7
С32—С33—Н33	121.6	С76—С77—Н77	121.7
C33—C34—C36	121.5 (8)	C77—C78—O12	127.9 (7)
C33—C34—O5	127.8 (7)	C77—C78—C80	122.3 (8)
C36—C34—O5	110.7 (7)	012	109.7 (7)
06-035-05	109.2 (7)	012	107.0 (6)
06—C35—H35A	109.8	012—C79—H79A	110.3
05-C35-H35A	109.8	011—C79—H79A	110.3
06—C35—H35B	109.8	012—C79—H79B	110.3
05-C35-H35B	109.8	011—C79—H79B	110.3
H35A-C35-H35B	108.3	H79A_C79_H79B	108.6
$C_{37} - C_{36} - C_{34}$	123.0 (8)	$C_{81} - C_{80} - O_{11}$	127.9 (8)
$C_{37} - C_{36} - O_{6}$	127.8 (7)	$C_{81} - C_{80} - C_{78}$	127.5 (8)
$C_{34} - C_{36} - O_{6}$	127.3(7) 109.1(7)	011 - C80 - C78	122.0(0) 109 5 (7)
$C_{36} = C_{37} = C_{31}$	117.8 (8)	$C_{80}$ $C_{81}$ $C_{75}$	107.5(7) 117.0(8)
$C_{30} = C_{37} = C_{31}$	121.1	$C_{80} = C_{81} = C_{75}$	117.0 (0)
$C_{30} = C_{37} = H_{37}$	121.1	$C_{30} = C_{31} = H_{31}$	121.5
$C_{31} = C_{37} = C_{37}$	121.1 116 5 (7)	$C^{7}$	121.3 117.0(7)
$C_{43} = C_{38} = C_{39}$	110.3(7) 121.0(7)	$C_{87} = C_{82} = C_{83}$	117.0(7)
$C_{43} = C_{58} = C_{20}$	121.9(7) 121.5(7)	$C_{8}^{2} = C_{82}^{2} = C_{70}^{2}$	121.0(7)
$C_{39} = C_{38} = C_{20}$	121.3(7)	$C_{83} = C_{82} = C_{70}$	121.4(7)
C40 - C39 - C38	122.0 (8)	$C_{84} = C_{83} = C_{82}$	121.9 (8)
C40-C39-H39	119.0	$C_{84} = C_{83} = H_{83}$	119.1
C38—C39—H39	119.0	C82—C83—H83	119.1
C39 - C40 - C41	119.7 (8)	C83 - C84 - C85	118.8 (8)
C39—C40—H40	120.1	C83—C84—H84	120.6
C41—C40—H40	120.1	C85—C84—H84	120.6
C40 - C41 - C42	119.9 (8)	C84 - C85 - C86	121.3 (8)
C40—C41—Br2	120.3 (6)	C84—C85—Br4	119.6 (6)
C42—C41—Br2	119.9 (6)	C86—C85—Br4	119.1 (6)
C43—C42—C41	119.1 (8)	C87/—C86—C85	119.5 (8)
C43—C42—H42	120.5	C87—C86—H86	120.3

C41—C42—H42	120.5	С85—С86—Н86	120.3
C42—C43—C38	122.7 (8)	C86—C87—C82	121.5 (8)
C42—C43—H43	118.6	С86—С87—Н87	119.3
C38—C43—H43	118.6	С82—С87—Н87	119.3
S2—C44—H44A	109.5	S4—C88—H88A	109.5
S2—C44—H44B	109.5	S4—C88—H88B	109.5
H44A—C44—H44B	109.5	H88A—C88—H88B	109.5
S2—C44—H44C	109.5	S4—C88—H88C	109.5
H44A—C44—H44C	109.5	H88A—C88—H88C	109.5
H44B—C44—H44C	109.5	H88B—C88—H88C	109.5
C22—S1—C1—C8	-80.2 (8)	C66—S3—C45—C52	-96.5 (8)
C22—S1—C1—C2	102.2 (7)	C66—S3—C45—C46	85.5 (7)
C8—C1—C2—C7	0.5 (9)	C52—C45—C46—C47	178.3 (9)
S1—C1—C2—C7	178.5 (6)	S3—C45—C46—C47	-3.4 (14)
C8—C1—C2—C3	176.4 (8)	C52—C45—C46—C51	0.4 (9)
S1—C1—C2—C3	-5.6 (13)	S3—C45—C46—C51	178.8 (6)
C7—C2—C3—C4	-2.1 (12)	C51—C46—C47—C48	-1.1 (12)
C1—C2—C3—C4	-177.5 (8)	C45—C46—C47—C48	-178.7 (8)
C2—C3—C4—C5	-1.0 (11)	C46—C47—C48—C49	-1.1 (11)
C2—C3—C4—C16	-178.6 (7)	C46—C47—C48—C60	-178.0(7)
C3—C4—C5—C6	3.3 (12)	C47—C48—C49—C50	2.1 (11)
C16—C4—C5—C6	-179.0 (7)	C60—C48—C49—C50	179.1 (7)
C4—C5—C6—C7	-2.3 (12)	C48—C49—C50—C51	-0.9 (12)
C5—C6—C7—O1	179.1 (7)	C52—O7—C51—C50	179.2 (8)
C5—C6—C7—C2	-1.0 (13)	C52—O7—C51—C46	0.6 (8)
C8—O1—C7—C6	179.9 (8)	C49—C50—C51—O7	-179.9 (7)
C8—O1—C7—C2	-0.1 (9)	C49—C50—C51—C46	-1.4 (12)
C3—C2—C7—C6	3.2 (13)	C47—C46—C51—O7	-178.9 (7)
C1—C2—C7—C6	179.8 (8)	C45—C46—C51—O7	-0.6 (9)
C3—C2—C7—O1	-176.8 (7)	C47—C46—C51—C50	2.4 (12)
C1—C2—C7—O1	-0.3 (9)	C45—C46—C51—C50	-179.3 (7)
C2-C1-C8-01	-0.6 (9)	C46—C45—C52—O7	-0.1 (9)
S1—C1—C8—O1	-178.5 (5)	S3—C45—C52—O7	-178.4 (5)
C2—C1—C8—C9	-177.9 (9)	C46—C45—C52—C53	-173.8 (9)
S1—C1—C8—C9	4.2 (14)	S3—C45—C52—C53	7.9 (15)
C7—O1—C8—C1	0.4 (9)	C51—O7—C52—C45	-0.3 (8)
C7—O1—C8—C9	178.4 (7)	C51—O7—C52—C53	175.2 (6)
C1—C8—C9—C10	-15.1 (15)	C45—C52—C53—C54	157.6 (9)
O1—C8—C9—C10	167.6 (7)	O7—C52—C53—C54	-16.0 (10)
C1—C8—C9—C15	166.8 (9)	C45—C52—C53—C59	-21.9(14)
O1—C8—C9—C15	-10.4(11)	O7—C52—C53—C59	164.4 (7)
C15—C9—C10—C11	-1.0(12)	C59—C53—C54—C55	-0.2(12)
C8—C9—C10—C11	-179.1 (8)	C52—C53—C54—C55	-179.7 (7)
C9—C10—C11—C13	-0.3(13)	$C_{53}$ $C_{54}$ $C_{55}$ $C_{56}$	0.7 (12)
C9 - C10 - C11 - O3	180 0 (7)	C54-C55-C56-O8	-1786(8)
C12—O3—C11—C10	178.9 (9)	C54—C55—C56—C58	-1.0(12)
C12-O3-C11-C13	-0.8(9)	C57—O8—C56—C55	-177.2(9)
$C_{11} = 0_{3} = C_{12} = 0_{2}^{2}$	16(9)	C57—O8—C56—C58	50(9)
$C_{13} - O_{2} - C_{12} - O_{3}$	-1.8 (9)	C58-09-C57-08	8.2 (10)

C10-C11-C13-C14	0.2 (13)	C56—O8—C57—O9	-8.1 (10)
O3—C11—C13—C14	179.9 (7)	C57—O9—C58—C59	176.1 (9)
C10-C11-C13-O2	180.0 (8)	C57—O9—C58—C56	-5.2 (9)
O3—C11—C13—O2	-0.3 (9)	C55—C56—C58—C59	0.8 (13)
C12—O2—C13—C14	-179.0 (9)	O8—C56—C58—C59	178.9 (8)
C12—O2—C13—C11	1.3 (9)	C55—C56—C58—O9	-178.0 (7)
C11—C13—C14—C15	1.3 (12)	O8—C56—C58—O9	0.1 (9)
O2—C13—C14—C15	-178.5 (8)	O9—C58—C59—C53	178.2 (7)
C13—C14—C15—C9	-2.6 (12)	C56—C58—C59—C53	-0.3 (12)
C10-C9-C15-C14	2.6 (12)	C54—C53—C59—C58	0.0 (12)
C8—C9—C15—C14	-179.3 (8)	C52—C53—C59—C58	179.5 (7)
C3—C4—C16—C21	19.7 (12)	C47—C48—C60—C65	22.7 (12)
C5-C4-C16-C21	-157.9 (7)	C49—C48—C60—C65	-154.2 (7)
C3—C4—C16—C17	-158.3 (8)	C47—C48—C60—C61	-158.1 (8)
C5-C4-C16-C17	24.1 (11)	C49—C48—C60—C61	25.0 (11)
C21—C16—C17—C18	4.9 (12)	C65—C60—C61—C62	2.9 (12)
C4—C16—C17—C18	-177.1 (8)	C48—C60—C61—C62	-176.3 (8)
C16—C17—C18—C19	-3.3 (13)	C60—C61—C62—C63	-1.8 (13)
C17—C18—C19—C20	-0.8 (13)	C61—C62—C63—C64	-0.5 (13)
C17—C18—C19—Br1	177.6 (6)	C61—C62—C63—Br3	177.1 (7)
C18—C19—C20—C21	3.2 (13)	C62—C63—C64—C65	1.6 (13)
Br1—C19—C20—C21	-175.2 (6)	Br3—C63—C64—C65	-176.0 (6)
C19—C20—C21—C16	-1.6(12)	C63—C64—C65—C60	-0.4(13)
C17—C16—C21—C20	-2.4(12)	C61—C60—C65—C64	-1.8(12)
C4-C16-C21-C20	179.5 (7)	C48—C60—C65—C64	177.4 (7)
C44—S2—C23—C30	90.9 (9)	C88—S4—C67—C74	-86.5(8)
C44 - S2 - C23 - C24	-912(7)	C88—S4—C67—C68	951(7)
C30-C23-C24-C25	-179.7(9)	C74—C67—C68—C69	173.8 (9)
<u>82—C23—C24—C25</u>	2.0 (14)	S4—C67—C68—C69	-7.5(13)
C30—C23—C24—C29	0.5 (9)	C74—C67—C68—C73	0.0 (9)
S2-C23-C24-C29	-177.8 (6)	S4—C67—C68—C73	178.7 (6)
C29—C24—C25—C26	0.3 (12)	C73—C68—C69—C70	-3.3(11)
$C_{23}$ $C_{24}$ $C_{25}$ $C_{26}$	-179.4(8)	C67—C68—C69—C70	-176.4(8)
$C_{24} = C_{25} = C_{26} = C_{27}$	12(12)	C68 - C69 - C70 - C71	20(11)
$C_{24} = C_{25} = C_{26} = C_{38}$	-1788(7)	C68 - C69 - C70 - C82	-1780(7)
$C_{25} - C_{26} - C_{27} - C_{28}$	-2.0(12)	C69 - C70 - C71 - C72	-0.3(12)
$C_{38} - C_{26} - C_{27} - C_{28}$	177 9 (8)	C82 - C70 - C71 - C72	179 8 (7)
$C_{26} = C_{27} = C_{28} = C_{29}$	1 3 (13)	C70-C71-C72-C73	-0.1(12)
$C_{20} = C_{21} = C_{22} = C_{23}$	-1786(8)	C74-010-C73-C68	2.2.(9)
$C_{30} - O_{4} - C_{29} - C_{24}$	0.8 (9)	C74 - 010 - C73 - C72	-1772(7)
$C_{27} - C_{28} - C_{29} - O_{4}$	179 8 (8)	C69 - C68 - C73 - O10	-1764(7)
$C_{27} = C_{28} = C_{29} = C_{24}$	0.4(12)	C67 - C68 - C73 - O10	-14(9)
$C_{25} = C_{24} = C_{29} = O_{4}$	179 3 (7)	C69 - C68 - C73 - C72	30(12)
$C_{23} - C_{24} - C_{29} - O_{4}$	-0.9(9)	C67 - C68 - C73 - C72	1780(7)
$C_{25} = C_{24} = C_{29} = C_{28}$	-12(13)	C71 - C72 - C73 - O10	178.1(7)
$C_{23}$ $C_{24}$ $C_{29}$ $C_{28}$	178 6 (7)	C71 - C72 - C73 - C68	-12(12)
$C_{24} = C_{23} = C_{30} = 04$	0.0 (9)	$C_{68} - C_{67} - C_{74} - O_{10}$	13(9)
S2-C23-C30-O4	178.1 (6)	S4—C67—C74—O10	-177 3 (6)
$C_{24}$ $C_{23}$ $C_{30}$ $C_{31}$	178.3 (9)	C68—C67—C74—C75	-179.0(9)
	(-)		

S2—C23—C30—C31	-3.5 (15)	S4—C67—C74—C75	2.5 (15)
C29—O4—C30—C23	-0.5 (9)	C73—O10—C74—C67	-2.1 (8)
C29—O4—C30—C31	-179.3 (7)	C73—O10—C74—C75	178.1 (6)
C23—C30—C31—C32	-173.2 (9)	C67—C74—C75—C76	165.1 (9)
O4—C30—C31—C32	5.2 (11)	O10-C74-C75-C76	-15.2 (11)
C23—C30—C31—C37	9.0 (15)	C67—C74—C75—C81	-14.0 (15)
O4—C30—C31—C37	-172.6 (7)	O10-C74-C75-C81	165.7 (7)
C37—C31—C32—C33	-0.4 (12)	C81—C75—C76—C77	-0.9 (12)
C30-C31-C32-C33	-178.3 (8)	C74—C75—C76—C77	180.0 (8)
C31—C32—C33—C34	-1.0 (13)	C75—C76—C77—C78	1.7 (13)
C32—C33—C34—C36	1.6 (13)	C76—C77—C78—O12	-177.4 (8)
C32—C33—C34—O5	177.8 (8)	C76—C77—C78—C80	-1.9 (13)
C35—O5—C34—C33	177.0 (9)	C79—O12—C78—C77	-173.5 (9)
C35—O5—C34—C36	-6.4 (9)	C79—O12—C78—C80	10.6 (9)
C36—O6—C35—O5	-8.3 (9)	C78—O12—C79—O11	-15.5 (9)
C34—O5—C35—O6	9.1 (9)	C80—O11—C79—O12	14.7 (9)
C33—C34—C36—C37	-1.0 (13)	C79—O11—C80—C81	172.8 (9)
O5—C34—C36—C37	-177.8 (7)	C79—O11—C80—C78	-8.4 (9)
C33—C34—C36—O6	178.2 (7)	C77—C78—C80—C81	1.3 (13)
O5—C34—C36—O6	1.4 (9)	O12—C78—C80—C81	177.5 (7)
C35—O6—C36—C37	-176.6 (9)	C77—C78—C80—O11	-177.6 (7)
C35—O6—C36—C34	4.3 (9)	O12-C78-C80-O11	-1.4 (9)
C34—C36—C37—C31	-0.4 (13)	O11—C80—C81—C75	178.3 (7)
O6—C36—C37—C31	-179.4 (7)	C78—C80—C81—C75	-0.3 (12)
C32—C31—C37—C36	1.1 (12)	C76—C75—C81—C80	0.2 (12)
C30—C31—C37—C36	178.9 (8)	C74—C75—C81—C80	179.3 (7)
C25—C26—C38—C43	160.9 (8)	C69—C70—C82—C87	24.2 (12)
C27—C26—C38—C43	-19.1 (12)	C71—C70—C82—C87	-155.9 (7)
C25—C26—C38—C39	-18.0 (12)	C69—C70—C82—C83	-157.2 (8)
C27—C26—C38—C39	162.0 (8)	C71—C70—C82—C83	22.7 (11)
C43—C38—C39—C40	2.1 (12)	C87—C82—C83—C84	2.2 (12)
C26—C38—C39—C40	-179.0 (7)	C70—C82—C83—C84	-176.5 (8)
C38—C39—C40—C41	0.0 (12)	C82—C83—C84—C85	1.3 (13)
C39—C40—C41—C42	-1.8 (12)	C83—C84—C85—C86	-3.9 (13)
C39—C40—C41—Br2	178.6 (6)	C83—C84—C85—Br4	175.5 (6)
C40—C41—C42—C43	1.3 (13)	C84—C85—C86—C87	3.0 (13)
Br2—C41—C42—C43	-179.0 (6)	Br4—C85—C86—C87	-176.5 (6)
C41—C42—C43—C38	0.9 (13)	C85—C86—C87—C82	0.7 (12)
C39—C38—C43—C42	-2.5 (12)	C83—C82—C87—C86	-3.2 (12)
C26—C38—C43—C42	178.5 (8)	C70—C82—C87—C86	175.5 (7)

Hye	drogen-bond geometry (2	Å,	?)		
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D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H···A
C12—H12A…O4	0.99	2.58	3.50(1)	154
C21—H21···O6 <sup>i</sup>	0.95	2.51	3.37 (1)	150
C57—H57A…O10 <sup>ii</sup>	0.99	2.50	3.32 (1)	141
C65—H65…O11 <sup>iii</sup>	0.95	2.58	3.47 (1)	156

C79—H79B···O7<sup>iv</sup> 0.99 2.56 3.26 (1) 128 Symmetry codes: (i) -x+2, -y+1, -z; (ii) -x+1, -y+1, -z+1; (iii) x, y+1, z; (iv) -x, -y+1, -z+1.



Fig. 1



Fig. 3



Molecule B 3.91Å 4.22Å Br2<sup>vi</sup> Br2<sup>vi</sup> Cg5 Cg6 Cg7 Cg8

